

Transformative Renewable Energy Storage Devices Based on Neutral Water Input

Energy Storage Systems Update
ARPA-E GRIDS Kick-Off
4 November 2010

Team

Proton Energy Systems



- Dr. Kathy Ayers, Pl
- Luke Dalton, System Lead
- Chris Capuano, Stack Lead
- Project Lead; Electrolysis Stack and System; Fuel
 Cell System
- Penn State University
 - Prof. Mike Hickner
 - Prof. Chao-Yang Wang
 - Electrolysis and Fuel Cell Membrane Material;
 Fuel Cell Stack





Proton Energy Systems

- Manufacturer of Proton Exchange Membrane (PEM) hydrogen generation products using electrolysis
- Founded in 1996
- Headquarters in Wallingford, Connecticut.
- ISO 9001:2008 registered
- Over 1,200 systems operating in 60 different countries





Certified to ISO 9001:2008



Precisely Right.





Proton Capabilities and Applications







PEM Cell Stacks

Complete Systems

Storage Solutions

- Complete product development, manufacturing & testing
- Containerization and hydrogen storage solutions
- Integration of electrolysis into RFC systems
- Turnkey product installation
- World-wide sales and service



Power Plants



Heat Treating



Semiconductors



Laboratories



Government



HOGEN® C Series

- Maximum Capacity: 30 Nm³/h H₂ (65 kg/day) (~200 kW input)
- Commercial availability: Q1 2011
- 5X hydrogen output with only 1.5X the foot print

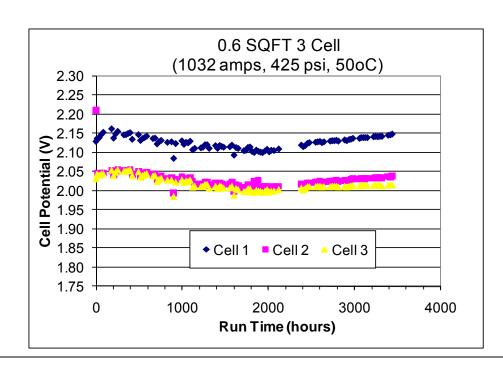


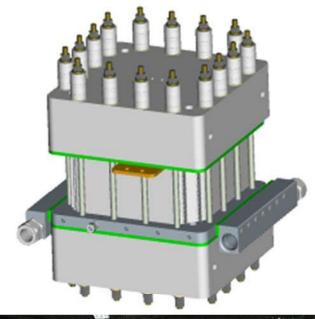




Next Steps in Scale Up

- 70 Nm³/h
- 150 kg/day
- 400 kW input

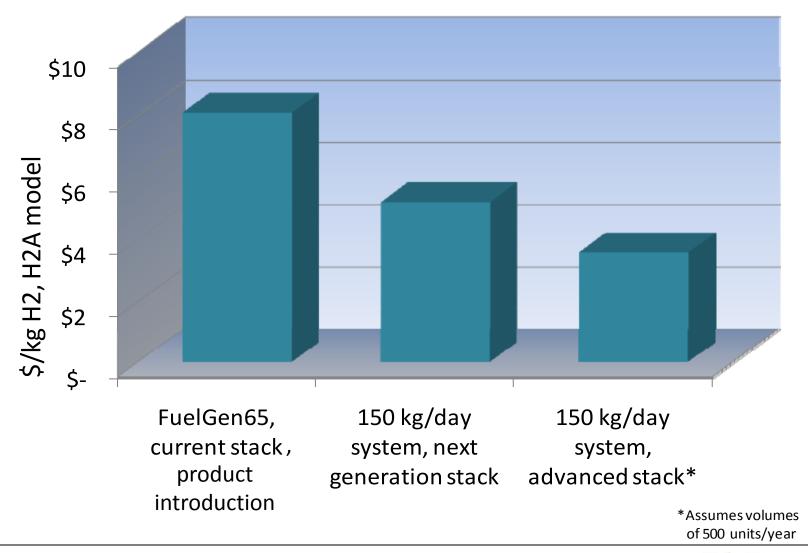








Hydrogen Cost Progression





Two Main Types of Low Temp Electrolysis

Alkaline Liquid Electrolyte

- Low bubble point requires balanced pressure
 - Controlled shutdown required
 - High pressure oxygen
- Corrosive solvent
- Complex balance of plant, high pressure lines
- Low current densities
- + Less expensive materials of construction

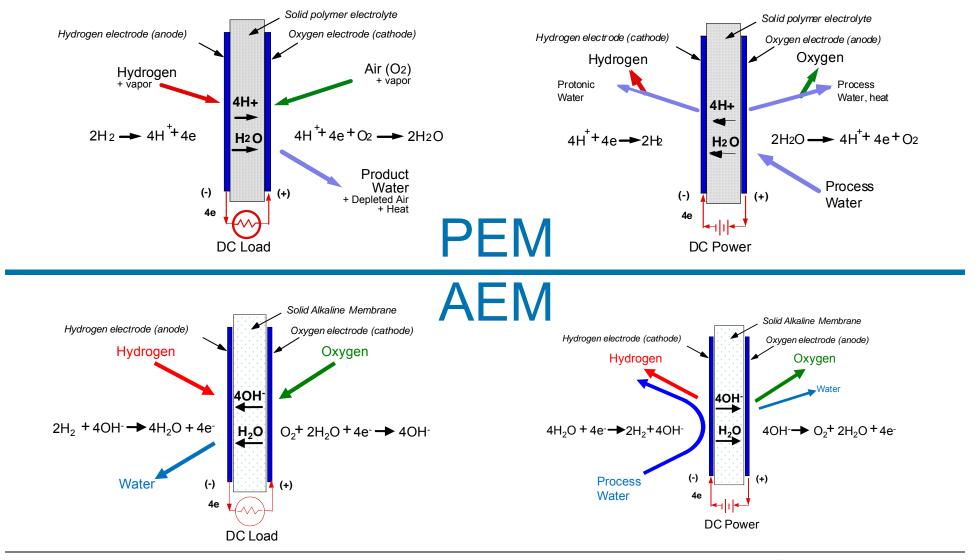
Proton Exchange Membrane (PEM)

- Membrane enables differential pressure
 - Load following
 - + Ambient pressure oxygen
- + Pure water
- Simple balance of plant,
 plastic on O₂ fluids loop
- + High current densities
- High cost catalysts and flow fields

Alkaline membrane technology could provide best of both systems



PEM / AEM Cell Comparison





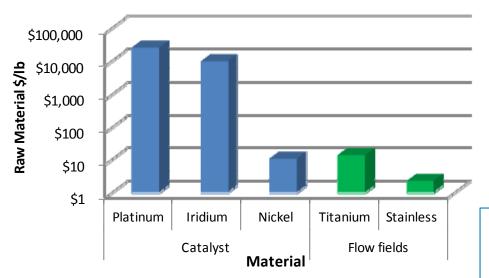
Rationale

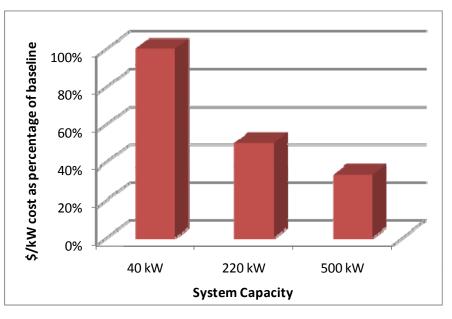
- Alkaline advantage over PEM: lower cost materials of construction
- Disadvantages of alkaline liquid system:
 - Corrosive electrolyte
 - High pressure oxygen
 - Complex balance of plant
 - Lower purity hydrogen
 - Lower efficiency
- Alkaline membranes showing feasibility
 - Enables PEM advantages at low cost
 - Enables lower current density for high efficiency



Cost Justification

3-pronged approach





Labor minimization and high speed manufacturing at high production volume



Proposed Approach

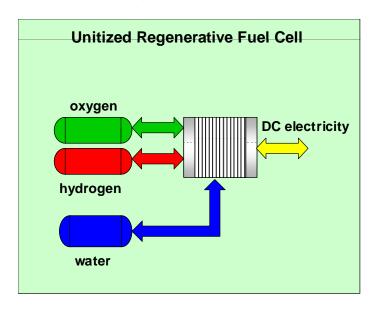
- Design Trade Study
 - Select RFC configuration
- Membrane and Ionomer Development
 - Maximize durability and minimize ionic resistance and crossover
- Catalyst Development
 - Reduce activation overpotential
- MEA Fabrication
 - Optimize catalyst-membrane interaction
- Cell Stack Design
 - Leverage Proton experience and substitute materials
- System Design
 - Leverage Proton balance of plant designs
- Cost Analysis



Trade Study:Configuration Options

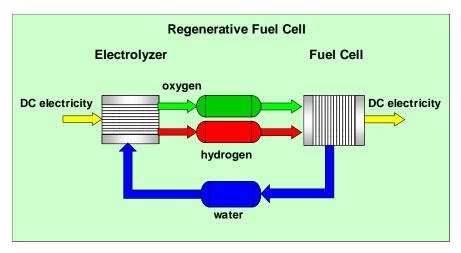
URFC – Unitized RFC

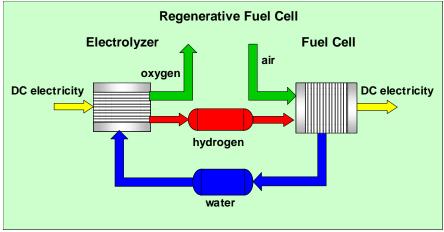
 A single cell stack that operates as both fuel cell and electrolyzer



DRFC – Discrete RFC

Separate fuel cell and electrolyzer stacks

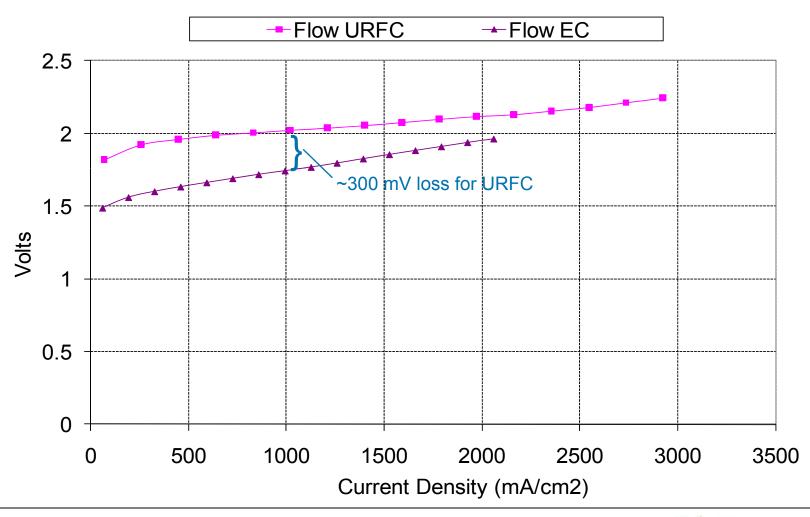






Trade Study – Electrolysis Example

Polarization Comparison





Trade Study: O₂ vs. air feed

- High pressure oxygen adds balance of plant complexity
- Requires special cleaning >150 psi
- O₂ feed requires drying of both gases

- Membrane is sensitive to CO₂
- Carbonate can replace OH- sites and reduce conductivity
- Removal of carbon from air reduces efficiency and increases cost

Need to look at trade of cost, efficiency, and simplicity in context of safety considerations



Trade Study: Anode vs. Cathode Feed

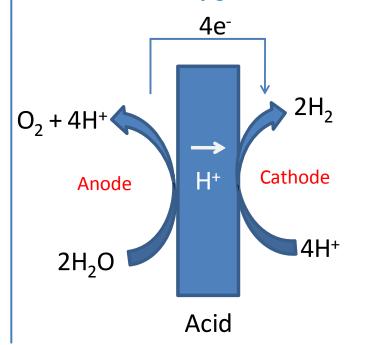
Alkaline electrolysis

- Water consumed on hydrogen side of cell
- Hydrogen primary interest

$O_2 + 2H_2O$ $O_2 + 2H_2O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_2 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_4 + 4H^+O$ $O_7 + 4H^+O$ $O_8 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_5 + 4H^+O$ $O_7 + 4H^+O$ $O_8 + 4H^+O$ $O_8 + 4H^+O$ $O_9 + 4H^+O$ $O_9 + 4H^+O$ $O_9 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_4 + 4H^+O$ $O_5 + 4H^+O$ $O_7 + 4H^+O$ $O_8 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_5 + 4H^+O$ $O_7 + 4H^+O$ $O_8 + 4H^+O$ $O_8 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_1 + 4H^+O$ $O_1 + 4H^+O$ $O_2 + 4H^+O$ $O_3 + 4H^+O$ $O_4 + 4H^+O$ $O_5 + 4H^+O$ $O_7 + 4H^+O$

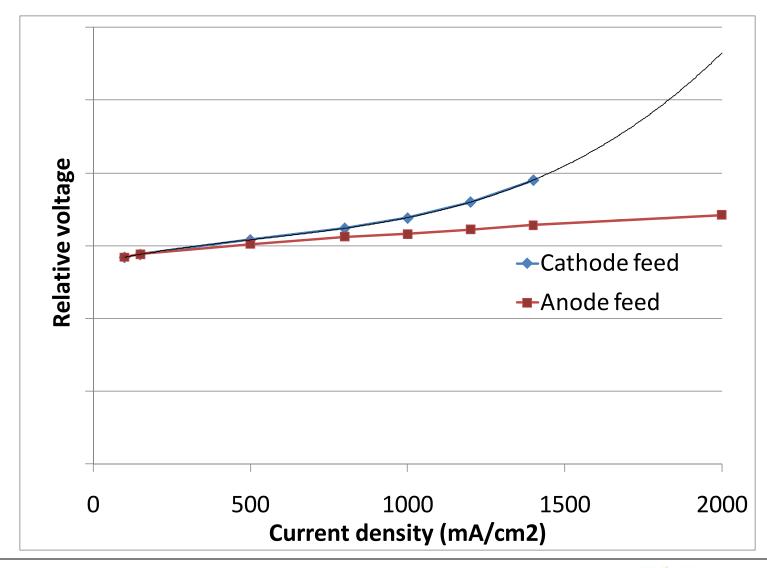
PEM electrolysis

- Water consumed on oxygen side of cell
- Hydrogen is typically gas of interest, oxygen vented





Trade Study: PEM Comparison





Membrane and Ionomer Development

- Penn State-led effort to develop an anion exchange membrane (AEM) and ionomer binder
 - Reduced cost through use of commercially available monomers
 - "Tune" ion exchange capacity and cross-linking to balance conductivity and mechanical properties/gas cross-over.
- Proton to assist with characterization
 - Conduct diffusion and performance evaluation to provide feedback on how to iterate on configuration
 - Have Tokuyama membranes and ionomers to test as a baseline for AEMs.



Catalyst Development

 Apply processes from PEM experience to AEM Surface area vs. catalyst synthesis and post treatment conditions 10.00 9.00 8.00 Normalized Surface Area vs. Baseline 7.00 6.00 5.00 4.00 3.00 2.00 1.00 0.00 atm 1, T1, atm 2, T2, atm 1, T2, atm 1, T2, atm 1, T2, atm 2, T3, atm 2 **Baseline** Catalyst



MEA Fabrication

- Vary temperature and dwell time
 - Find combination most conducive to electrode attachment
- Sub-scale membrane samples to be used for pressing half-MEAs
 - Two fabrication approaches to be considered
- Trials rated on uniformity, adhesion to membrane, and degree of flow for ink samples.
- Membrane assessed for mechanical degradation resulting from process

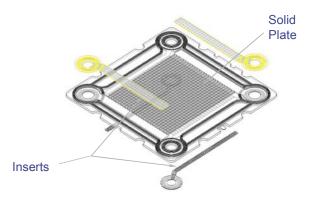


Electrolysis Stack Design

- Will incorporate outputs from the trade study to dictate configuration
 - Bipolar plate design (high-efficiency, low-pressure)
 - Round design (lower-efficiency, high pressure)
- Primary effort will be material substitution

 Alkaline allows replacement of titanium components with stainless steel.

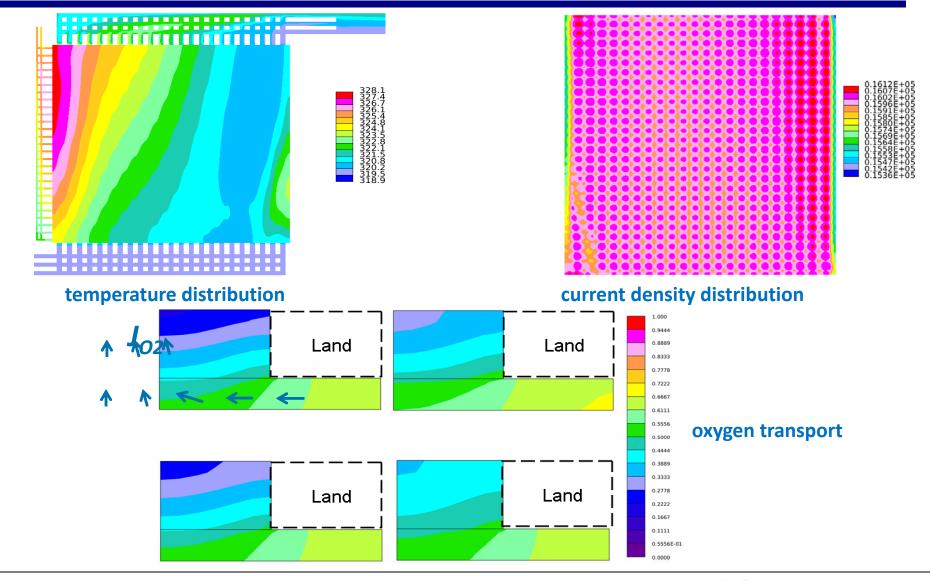




Bipolar plate design

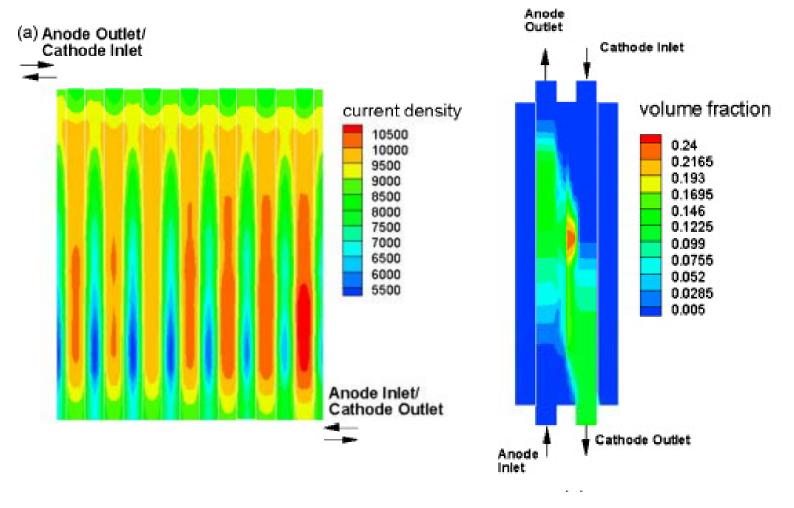


Electrolysis Cell Modeling





Fuel Cell Modeling



current density distribution

Liquid water distribution



Electrolysis and Fuel Cell Stack Test Plan

- Design phase and concept review
- Prototype flow field fabricated using production tooling and techniques
- Anode flow field verification
- Cathode flow field verification
- Short stack testing and operation for prototype review
- Deliverable stack assembly and operation



Cell Stack Operational Verification

- Multi-cell operational testing
- Allows for fine control over operating parameters
 - Temperature
 - Generation Pressure
 - Current Control
 - Water Flow Rates





Energy Storage System Design

- PEM system should be largely transferrable to proposed AEM-based system
 - Common fluids of interest (H₂, H₂O, O₂)
- Trade study work will impact type of system
- Leverage prior experience in closed loop REFCs
- Output:
 - Design Intent Document
 - P&ID
 - Component selection (comprehensive BOM)
 - Haz-op and design review



Alkaline REFC Approach

Well-suited to load-following

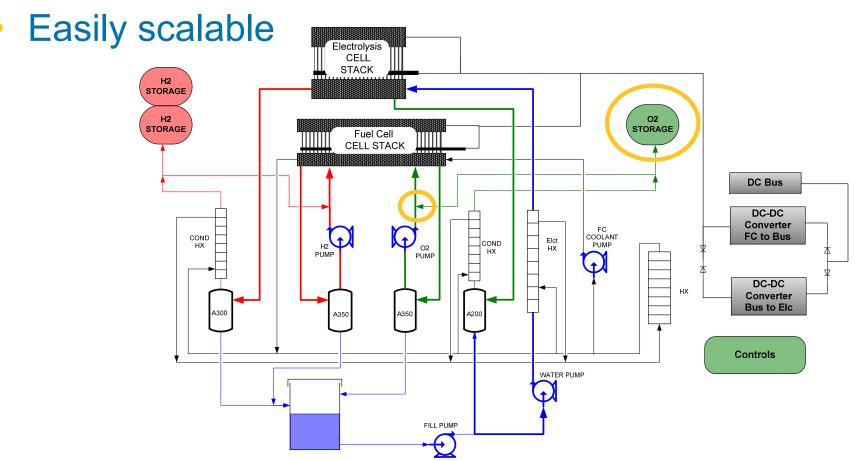


Figure 1. Closed-loop, pure oxygen, discrete regenerative fuel cell system P&ID.



Examples of Demonstrated Energy Storage Systems



Missile Defense Agency: Closed loop RFC

"Regenerative Fuel Cell" integrated into CERL's Silent Camp system concept







Near-Term Milestones

- Kickoff meeting October 6, 2010
- Trade study results on discrete vs. unitized stack
- Initial survey of commercial catalysts
- MEA formulation studies with baseline membrane and catalyst
- Survey of stack part availability in alternate materials



Major Project Milestones

- System Configuration Trade Study
- Electrolysis and Fuel Cell Membrane Material Improvements
- Electrolysis and Fuel Cell Membrane-Electrode-Assembly Fabrication Optimization
- Fuel Cell Stack Design and Test
- Electrolysis Stack Design and Test
- Integrated Energy Storage System Demonstration

